The Biosynthesis of Glaucine in Litsea glutinosa

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Incorporation of ¹⁴C labelled (\pm) -reticuline and tritiated racemic reticuline, protosinomenine, orientaline, laudanosine, isoboldine, boldine, thaliporphine, *N*-methyl-laurotetanine, and dicentrine into glaucine in young *Litsea glutinosa* (Lour) C. B. Rob. var. *glabraria* Hook f. (Lauraceae) has been studied. The results provide evidence that (S)-reticuline (5) is converted by oxidative coupling into (S)-isoboldine (6) and methylation *via* (S)-thaliporphine (7) to form (S)-glaucine (8).

Glaucine (8),¹ the antithrombotic, anti-inflammatory, and analgesic principle of several plant species ² has been isolated both in (+)- and (-)-forms. The structure and stereochemistry of (+)-glaucine (8)³ is well established.⁴ (-)-Glaucine (17) has been isolated from Korean *Corydalis* tubers.⁵

Glaucine (8) could be formed in Nature from reticuline (1), protosinomenine (2), and orientaline (3) by alternative biosynthetic routes.⁶ However, it has been reported that glaucine (8) in *Dicentra eximia* (Ker). Toer (Fumariaceae) is biosynthesized from norprotosinomenine.⁷ We have studied the biosynthesis of several aporphine alkaloids⁸⁻¹⁰ and we now find that (8) is biosynthesized from (S)-reticuline (5) in *Litsea glutinosa* (Lour) C. B. Rob. var. *glabraria* Hook f. (Lauraceae).

Initially reticuline (1) (experiment 2), protosinomenine (2) (experiment 3), and orientaline (3) (experiment 4) were fed in parallel with tyrosine (experiment 1) to young *Litsea glutinosa*

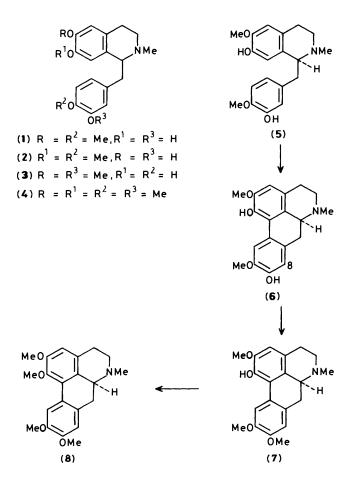


Table. Tracer experiments on Litsea glutinosa

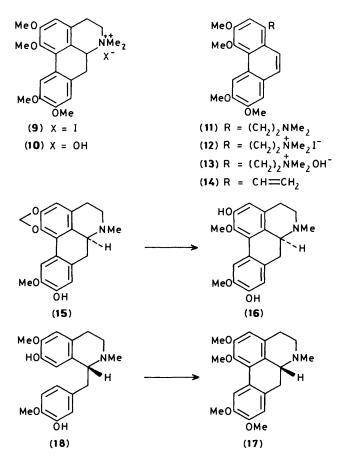
Expt.	Precursor fed	% Incorporation into Glaucine (8)	
1	(L)-[U- ¹⁴ C]Tyrosine	0.02	0.02
2	(\pm) -[2',6',8- ³ H ₃]Reticuline (1)		0.14
3	(\pm) -[Aryl- ³ H]Protosinomenine (2)		0.04
4	(\pm) -[Aryl- ³ H]Orientaline (3)		0.02
5	(\pm) -[Aryl- ³ H]Laudanosine (4)		0.002
6	(\pm) -[3- ¹⁴ C]Reticuline (1)		0.16
7	$(\pm)-[3,8-{}^{3}H_{2}]Boldine (16)$		0.06
8	(\pm) -[Aryl- ³ H]Isoboldine (6)		0.21
9	(+)-[Aryl- ³ H]Thaliporphine (7)		0.27
10	(\pm) -[Aryl- ³ H]-N-Methyl-laurotetanine		0.06
11	$(+)-[Aryl-^{3}H]$ Dicentrine (15)		0.01
12	$(S)-(+)-[2',6',8-^{3}H_{3}]$ Reticuline (5)		0.18
13	$(R)-(-)-[2',6',8-^{3}H_{3}]$ Reticuline (18)		0.012

var. glabraria plants and it was found that reticuline (1) was incorporated with higher efficiency into glaucine (8) than protosinomenine (2), and orientaline (3). The incorporation of (2) and (3) was low but not negligible. As expected, laudanosine (4) (experiment 5) was not incorporated into (8). The incorporation of various labelled precursors into glaucine is recorded in the Table.

The regiospecificity of the label in the biosynthetic glaucine (8) derived from (\pm) -[3-¹⁴C]reticuline (1) (experiment 6) was determined as follows. Biosynthetic glaucine (8) was treated with methyl iodide to afford glaucine methiodide (9) having essentially the same molar radioactivity as the parent base. Radioactive (9) was converted into the corresponding methohydroxide (10) by IR-410 anion exchange resin. Hofmann degradation of compound (10) gave the methine-I (11) with essentially no loss of radioactivity. Treatment of (11) with methyl iodide afforded glaucine metholydroxide (13). Second Hofmann degradation of the labelled compound (13) yielded radioactive 3,4,6,7-tetramethoxy-1-vinylphenanthrene (14). Ozonolysis of compound (14) gave a radioactive formaldehyde dimethone derivative (85% of original activity).

Although the foregoing experiments established that reticuline (1) was specifically incorporated into glaucine (8) in *L.* glutinosa var. glabraria, the precursors used were racemic. Parallel feedings of labelled (+)-reticuline (5) (experiment 12) and (-)-reticuline (18) (experiment 13) demonstrated that stereospecificity is maintained in the bioconversion of (5) into (8) in *L. glutinosa* var. glabraria plants. The former was incorporated about 10 times more efficiently than the latter. The incorporation of (-)-reticuline into glaucine (8) is due to the optical impurity of (+)-reticuline (5).

Specific incorporation of (S)-reticuline (5) into glaucine (8) implied isoboldine (6) as an intermediate. Labelled (6)



(experiment 8) when fed to young L. glutinosa var. glabraria plants was efficiently incorporated into (8). Labelled boldine (16) (experiment 7) when fed to young L. glutinosa var. glabraria plants gave rise to radioactive biosynthetic (8), suggesting thereby that the enzymes present in the plant can metabolize boldine (16) to give (8).

In the bioconversion of isoboldine (6) into glaucine (8), selective O-methylation at position 1 in isoboldine (6) could give N-methyl-laurotetanine. Alternatively selective O-methylation at position 9 in isoboldine (6) could give thaliporphine (7). Glaucine (8) could then be formed finally both from N-methyllaurotetanine and from (7) by O-methylation at positions 9 and 1, respectively. Labelled thaliporphine (7) (experiment 9), when fed in parallel with labelled N-methyl-laurotetanine (experiment 10) to young L. glutinosa var. glabraria plants, was incorporated into (8) about five times more efficiently than Nmethyl-laurotetanine demonstrating thereby that selective Omethylation at position 9 in isoboldine (6) occurs to give thaliporphine (7) which is then converted into (8). Labelled dicentrine (15) (experiment 11) when fed to young L. glutinosa var. glabraria plants was poorly metabolized to give (8).

(S)-Reticuline (5) and (S)-isoboldine (6) have been isolated from L. glutinosa var. glabraria.¹¹ The presence of these bases in young L. glutinosa var. glabraria plants was confirmed by a trapping experiment with (L)-[U-1⁴C]tyrosine (incorporation 0.3 and 0.2% respectively). (S)-Reticuline (5) and isoboldine (6) are, thus, the biological precursors of glaucine (8) in young L. glutinosa var. glabraria plants. The foregoing tracer experiments support the following sequence for the biosynthesis of glaucine (8) in young L. glutinosa (Lour.) C. B. Rob. var. glabraria Hook f. (Lauraceae): Tyrosine $\longrightarrow (S)-(+)$ -reticuline (5) $\longrightarrow (S)$ isoboldine (6) $\longrightarrow (S)$ -thaliporphine (7) $\longrightarrow (S)$ -glaucine (8). Several aporphine,^{8-10,12} protoberberine, and tetrahydroprotoberberine $^{13-18}$ alkaloids, in recent years have shown to be derived in Nature from reticuline. Although in the majority of the cases the orientation of hydroxy and methoxy functions in the isoquinoline moiety of the precursor and the corresponding position in the biosynthetic bases is the same, in a few cases the orientation of these groups is different.¹⁰ It has been shown that the methylenedioxy function in berberine is formed from *ortho*hydroxy methoxy function.^{19,20} It is possible that dicentrine (**15**) could be an intermediate in the biosynthesis of boldine (**16**) from isoboldine (**6**).

Glaucine has been isolated both in (+)- and (-)-forms from plants. We have demonstrated with the help of tracer experiments that (S)-glaucine (8) is stereospecifically derived in L. glutinosa var. glabraria plants from (S)-reticuline (5). It is logical that (R)-glaucine (17) is also stereospecifically derived in plants from (R)-reticuline (18).

Experimental

For general directions (spectroscopy details and counting method) see ref. 21.

Synthesis of Precursors.—The racemates of reticuline (1),²² protosinomenine (2),²³ orientaline (3),²⁴ and laudanosine (4)¹⁰ were prepared by standard procedures. (+)-Isoboldine (6), (+)-N-methyl-laurotetanine, (+)-dicentrine (15), and (±)-thaliporphine (7) used, were natural products.²⁵

Resolution.— (\pm) -Di-O-benzylreticuline was resolved ²⁶ by treatment with (+)- and (-)-O,O-dibenzoyltartaric acids. Hydrogenolysis of the benzyl ethers with HCl furnished (S)-(+)- and (R)-(-)-reticulines, (5) and (18) respectively, of known absolute configuration.

Labelling of Precursors: Tritiation.—Both acid- and basecatalysed exchange reactions ^{27,28} were employed.

 (\pm) -[2',6',8-³H₃]*Reticuline*. (\pm)-Reticuline (100 mg) in tritiated water (0.2 ml, 200 mCi) containing Bu'OK (220 mg) was heated under N₂ (sealed tube) for 110 h at 100 °C. Work-up afforded (\pm)-[2',6',8-³H₃]reticuline. (S)-(+)- and (R)-(-)-[2',6',8-³H₃] Reticulines, (5) and (18) respectively, and (+)-[3,8-³H₂]boldine (16) were prepared in the same way.

(\pm)-[Aryl-³H]*Protosinomenine* (**2**). Compound (**2**) (110 mg) was added to a mixture of tritiated water (0.2 ml, 200 mCi) and SOCl₂ (0.1 ml), and the mixture under N₂ (sealed tube) was heated for 110 h at 100 °C. The resulting mixture was worked up to give (\pm)-[*aryl*-³H]protosinomenine (**2**). (\pm)-Orientaline (**3**), (\pm)-laudanosine (**4**), isoboldine (**6**), thaliporphine (**7**), *N*-methyl-laurotetanine, and dicentrine (**15**) were tritiated in the same way.

 (\pm) -[3-¹⁴C]Reticuline (1) was prepared by total synthesis.

Feeding Experiments.—For feeding purposes, the labelled precursors were either dissolved in water (1 ml) containing tartaric acid (10 mg) or in aqueous dimethyl sulphoxide (0.2 ml). Twigs of *Litsea glutinosa* (Lour.) C. B. Rob. var. glabraria Hook f. (Lauraceae) were dipped into the solution of the precursor. When uptake was complete, the twigs were washed with water and worked up to provide biosynthetic glaucine (8).

Isolation of Glaucine (8).—The precursor-fed twigs (typically 40 g, wet wt.) of L. glutinosa var. glabraria plants were macerated in EtOH (250 ml) with radioinactive glaucine (8) (100 mg) and left for 10 h. The EtOH was decanted and the plant material was percolated with fresh EtOH (6×250 ml). The combined washings were concentrated under reduced pressure to afford a greenish viscous mass which was extracted with 5% hydrochloric acid (5×20 ml). The acidic extract was defatted

with light petroleum $(3 \times 20 \text{ ml})$, basified with Na₂CO₃ (pH 8), and extracted with CHCl₃ (6 × 30 ml). The combined CHCl₃ extracts were washed with water, dried (Na₂SO₄), and evaporated. The crude base was subjected to preparative t.l.c. on silica gel GF₂₅₄ (solvent: CHCl₃–MeOH, 97:3). The region containing glaucine (8) was removed and eluted with CHCl₃–MeOH (80:20). The solvent from the eluate was removed under reduced pressure to give radioactive glaucine (8) (78 mg), m.p. 120–121 °C (from MeOH–H₂O) (lit.,²⁹ 120 °C).

Degradation of the Biosynthetic Glaucine (8) Derived from (\pm) -[3-¹⁴C]*Reticuline.*—Labelled glaucine (8) (250 mg) (molar activity 6.60×10^4 disint. min⁻¹ mmol⁻¹) in MeOH (10 ml) was heated with MeI (5 ml) for 2 h to give radioactive glaucine methiodide (9) (275 mg), m.p. 222–223 °C (lit., ³⁰ 221 °C) (molar activity 6.62×10^4 disint. min⁻¹ mmol⁻¹). A solution of (9) in MeOH (60 ml) was passed through a column of freshly generated Amberlite IR-410 anion exchange resin (OH⁻ form) to afford the corresponding radioactive methohydroxide (10). Labelled (10) in MeOH (15 ml) was refluxed with KOH (4 g in 4 ml water) for 2.5 h. The solvent from the resulting mixture was removed under reduced pressure, and the residue diluted with water (10 ml) and extracted with $Et_2O-CHCl_3$ (3:1; 5 × 30 ml). The combined extracts were washed with water, dried (Na_2SO_4) , and evaporated to give radioactive glaucine methylmethine (11) (190 mg) as an oil.

Labelled (11) (188 mg) in MeOH (5 ml) was refluxed with MeI for 2 h to give radioactive glaucine methylmethine methiodide (12) (184 mg), m.p. 278-280 °C (lit., ³⁰ 276-280 °C) (molar activity 6.48×10^4 disint. min⁻¹ mmol⁻¹). Labelled (12) was converted into the corresponding methohydroxide (13) by passage through Amberlite IR-410 anion exchange resin (OH⁻ form). Labelled (13) in MeOH (10 ml) was refluxed with KOH (3.5 g) in water (3 ml) for 2 h. The solvent from the resulting mixture was removed under reduced pressure and the concentrate was diluted with water and extracted with $Et_2O-CHCl_3$ (3:1; 4 × 30 ml). The combined extracts were washed with water, dried (Na₂SO₄), and concentrated under reduced pressure to afford radioactive 3,4,6,7tetramethoxy-1-vinylphenanthrene (14) (102 mg), m.p. 142-143 °C (lit..³⁰ 143 °C) (molar activity 6.00×10^4 disint. min⁻¹ $mmol^{-1}$).

Ozonised O₂ was passed through a solution of radioactive (14) (100 mg) in EtOAc (8 ml) at -78 °C for 20 min. The solvent from the resulting mixture was removed under reduced pressure and water, Zn dust (320 mg), and AgNO₃ (15 mg) were added to the residue. The mixture was refluxed for 20 min and distilled. The distillate was collected in a solution of dimedone (310 mg) in aqueous EtOH (80 ml) which after being kept for 1 h, was concentrated to 10 ml and left overnight. The precipitate was chromatographed over silica gel column. Elution with C₆H₆-CHCl₃ and CHCl₃ (t.l.c. control) furnished labelled formaldehyde dimethone, m.p. 187–188 °C (lit.,³¹ m.p. 188 °C) (molar activity 5.1 × 10⁴ disint. min⁻¹ mmol⁻¹; 85% of original activity).

Trapping Experiment.—(L)-[U-1⁴C]Tyrosine (activity 0.5 mCi) was fed to the twigs of *L. glutinosa* var. glabraria plant and after 7 days the twigs were harvested. The plant material (90 g, wet wt.) was macerated in EtOH (250 ml) with inactive reticuline (5) (100 mg) and isoboldine (6) (48 mg) separately each being left for 12 h. The EtOH was decanted and the plant material percolated with fresh EtOH (5 × 200 ml). The combined washings were concentrated under reduced pressure

to afford a greenish viscous mass which was extracted with 5% HCl (4 × 15 ml). The acidic extract was defatted with light petroleum (3 × 10 ml), basified with Na₂CO₃ (pH 8), and extracted with CHCl₃ (5 × 15 ml). The combined extracts were washed with water, dried (Na₂SO₄), and concentrated under reduced pressure to furnish the crude alkaloidal mixture which was subjected to preparative t.l.c. on silica gel GF₂₅₄ (solvent: CHCl₃-MeOH, 90:10) to afford radioactive (5), base picrate, m.p. 189–190 °C (lit.,³² 190–192 °C, incorporation 0.3%) and (6), m.p. 121–122 °C (lit.,³³ 122–123 °C, incorporation 0.2%).

References

- 1 R. H. F. Manske, 'The Alkaloids,' R. H. F. Manske and H. L. Holmes (eds.), Academic Press, New York, 1954, vol. IV, p. 120.
- 2 H. Guinaudeau, M. Leboeuf, and A. Cave, Lloydia, 1975, 38, 275.
- 3 T. A. Henry, 'The Plant Alkaloids,' J. and A. Churchill Ltd., London, 1949, p. 311.
- 4 Probst, Justus Liebigs Ann. Chem., 1839, 31, 241.
- 5 J. Go, J. Pharm. Soc. Jpn., 1930, 50, 933.
- 6 A. R. Battersby, in 'Oxidative Coupling of Phenols,' A. R. Battersby and W. I. Taylor (eds.), Dekker, New York, 1967, p. 119.
- 7 A. R. Battersby, J. L. McHugh, J. Staunton, and M. Todd, J. Chem. Soc., Chem. Commun., 1971, 985.
- 8 O. Prakash, D. S. Bhakuni, and R. S. Kapil, J. Chem. Soc., Perkin Trans. 1, 1978, 622.
- 9 D. S. Bhakuni, S. Jain, and R. S. Singh, *Tetrahedron*, 1980, **36**, 2525. 10 D. S. Bhakuni, S. Tewari, and R. S. Kapil, *J. Chem. Soc.*, *Perkin*
- *Trans.* 1, 1977, 706. 11 N. K. Hart, S. R. Johns, J. A. Lamberton, J. W. Loder, A. Moorhouse,
- 11 N. K. Hart, S. R. Johns, J. A. Lamberton, J. W. Loder, A. Moornouse, A. A. Sioumis, and T. K. Smith, Aust. J. Chem., 1969, 22, 2259.
- 12 G. Blaschke, Arch. Pharm., 1970, 303, 358.
- 13 A. R. Battersby, R. J. Francis, M. Hirst, and J. Staunton, Proc. Chem. Soc., 1963, 268.
- 14 D. H. R. Barton, R. H. Hesse, and G. W. Kirby, J. Chem. Soc., 1965, 6379.
- 15 D. S. Bhakuni, S. Jain, and S. Gupta, Tetrahedron, 1980, 36, 2491.
- 16 D. S. Bhakuni, S. Jain, and S. Gupta, Tetrahedron, 1983, 39, 455.
- 17 A. R. Battersby, R. J. Francis, E. A. Ruveda, and J. Staunton, J. Chem. Soc., Chem. Commun., 1965, 89.
- 18 D. S. Bhakuni, S. Gupta, and S. Jain, Tetrahedron, 1983, 39, 4003.
- 19 C. W. W. Beecher and W. J. Kelleher, Tetrahedron Lett., 1983, 24, 469.
- 20 M. Rueffer, O. Ekundayo, N. Nagakura, and M. H. Zenk, *Tetrahedron Lett.*, 1983, 24, 2643.
- 21 D. S. Bhakuni, S. Jain, and A. N. Singh, J. Chem. Soc., Perkin Trans. 1, 1978, 380 and ref. cited.
- 22 D. H. R. Barton, R. James, G. W. Kirby, D. W. Turner, and D. A. Widdowson, J. Chem. Soc. C, 1968, 1529.
- 23 S. Teitel, J. O. Brien, and A. Brossi, J. Med. Chem., 1972, 15, 845.
- 24 M. Tomita and J. Kunitomo, J. Pharm. Soc. Jpn., 1960, 80, 1238.
- 25 S. Tewari, Ph.D. Thesis, Lucknow University, 1974.
- 26 A. R. Battersby, D. M. Foulkes, and R. Binks, J. Chem. Soc., 1965, 3323.
- 27 G. D. Rittenberg, A. S. Keston, R. Schoenheimer, and G. L. Fosler, J. Biol. Chem., 1938, 125.
- 28 C. K. Ingold, C. G. Raisin, and C. L. Wilson, J. Chem. Soc., 1936, 1637.
- 29 M. Tomita and J. Furukawa, J. Pharm. Soc. Jpn., 1962, 82, 1199.
- 30 M. Tomita and F. Fusada, Chem. Pharm. Bull., 1953, 1, 5.
- 31 A. R. Battersby and B. J. T. Harper, J. Chem. Soc., 1962, 3526.
- 32 M. K. Jain, J. Chem. Soc., 1962, 2203.
- 33 S. R. Johns, J. A. Lamberton, and A. A. Sioumis, Aust. J. Chem., 1966, 19, 2331.

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